A SYNTHESIS OF CIS-JASMONE

Steven M. Weinreb and Raymond J. Cvetovich¹
Department of Chemistry, Fordham University
Bronx, New York 10458

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cis-Jasmone (6), an important constituent of the essential oil of jasmine, has been the object of a number of synthetic efforts in recent years. 2-6 The more successful approaches have involved cyclization of an appropriately substitued 1,4-diketone to the jasmone cyclopentenone system. 2-6 Although several ingenious new methods for construction of 1,4-diketones have been reported, 2,5,6,8 one of the best routes to this system (Scheme I) involves lithiation of a furan, followed by alkylation and hydrolysis. The major disadvantage of this route is the rather vigorous hydrolytic conditions necessary for opening the furan ring, sometimes resulting in undesired changes in the side chain. It appeared that a hydroxycyclobutanone might serve as the synthetic equivalent of 2-methylfuran, providing an alternative 1,4-diketone synthon.

2-Hydroxy-2-methylcyclobutanone (1), readily available in quantitative yield by irradiation of 2,3-pentanedione, 9,10 was converted in 75% yield to a mixture of diastereomeric acetals 2 [ν_{max} film 1780 cm $^{-1}$] upon treatment with ethyl vinyl ether

Scheme I

in tetrahydrofuran solution in the presence of Dowex cation exchange resin. Addition of an ether solution of this mixture of acetals2 to the Grignard reagent 3, prepared in ether from the corresponding bromide, produced a diastereoisomeric mixture of alcohols 4. Without purification, this mixture was stirred overnight with periodic acid in aqueous tetrahydrofuran, giving the known diketone 5 (35% from 2). It had infrared and NMR spectra as described by Buchi and Crombie to $\nu_{\rm max}$ CHCl₃ 1712, 1400 and 1360 cm⁻¹; NMR (CDCl₃) δ 5.4 (2H, m) 2.70 (4H, s), 2.5-1.5 (6H, m), 2.18 (3H, s), 0.98 (3H, s, J = 7.5 Hz).

Cyclization of diketone 5 with dilute sodium hydroxide 3,4,7 produced cisjasmone (6) having infrared, ultraviolet and nuclear magnetic resonance spectra as reported. 3-5,11

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